Modeling and Simulation in Chemical Engineering Question bank Part-I Department of Chemical Engineering

- 1. What is the mathematical model?
- Write the names of software's available for mathematical modelling and simulation for chemical engineering applications. Discuss how they help and differ from each other in solving the problems.
- 3. Classify the mathematical model
- 4. Explain why modelling assumptions are important in building of a 'model'?
- 5. One of your friends writes down the inventory rate equation for money as

$$\begin{pmatrix} Change \text{ in amount} \\ of \text{ rupees} \end{pmatrix} = (Interest) - \begin{pmatrix} Service \\ charge \end{pmatrix} + \begin{pmatrix} Rupees \\ deposited \end{pmatrix} - \begin{pmatrix} Checks \\ written \end{pmatrix}$$

Identify the terms in the above equation

- Determine whether steady or unsteady state conditions prevail for the following cases (a). The height of water in dam during heavy rain (b). The weight of an athlete during a marathon (c). The temperature of an ice cube as it melts.
- 7. How to determine the rate of quantity interms of (i) Inlet/outlet rate (ii) Generation rate (iii) Accumulation rate.
- 8. Steady state transport without generation
- 9. Steady state transport with generation
- 10. Derive the total flux interms of Peclet number
- 11. Express (or) derive the mass flow rate of species 'i' entering or leaving the system in a mathematical form.
- 12. Explain the term "Peclet number" and its significance.
- 13. Recall the generalized form of molecular and convective flux of conserved quantities
- 14. The differential equation describing the time rate of change of water height is given by $\frac{dh}{dt} = 6 8\sqrt{h}$ Where, h is the height of water in meters. Calculate the height of water in the tank under steady conditions.
- 15. Dust evolves at a rate of 0.3 kg/ h in a foundry which has the dimensions of 20 m × 8 m × 4 m. According to ILO (International Labor Organization) standards, the dust concentration should not exceed 20 mg/ m³ to protect workers health. Determine the volumetric flow rate of ventilating air to meet the standards of ILO.

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- 16. Steam at a temperature of 200 °C flows through a pipe of 5 cm inside diameter and 6cm outside diameter. The length of the pipe is 30m. If the steady rate of heat loss per unit length of the pipe is 2W/m, calculate the heat fluxes at the inner and outer surfaces of the pipe.
- 17. Air at atmospheric pressure and 95 °C flows at 20 m/s over a flat plate of naphthalene 80 cm long in the direction of flow and 60 cm wide. Experimental measurements report the molar concentration of naphthalene in the air, C_A, as a function of distance *x* from the plate as follows:

<i>x</i> , (cm)	0	5	15	25	35	45
$C_{\rm A}$, (mol/m ³)	0.117	0.105	0.085	0.069	0.057	0.047

Determine the molar flux of naphthalene from the plate surface under steady conditions by using least squares method. (Data: Diffusion coefficient of naphthalene (A) in air (B) at 25 $^{\circ}C = 0.62 \times 10^{-5} \text{ m}^2/\text{s}$).

- 18. The geothermal gradient is the rate of increase of temperature with depth in the earth's crust.
 a) If the average geothermal gradient of the earth is about 25 °C/ km, estimate the steady rate of heat loss from the surface of the earth. Take the diameter and the thermal conductivity of the earth as 1.27 x 10⁴ km and 3 W/ (m. K), respectively.
- 19. Air at 20 °C and 1 atm pressure flows over a porous plate that is soaked in ethanol. The molar concentration of ethanol in the air, C_A , is given by $C_A = 4 e^{-1.5z}$. Where, C_A is in kmol/m³ and z is the distance measured from the surface of the plate in meters. Calculate the molar flux of ethanol from the plate. Diffusivity of Ethanol in air is 1.1 x 10⁻⁶ m²/s.